

## ISOLATION AND STRUCTURAL STUDIES ON THE ALKALOIDS OF *PETCHIA CEYLANICA*

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**Key Word Index**—*Petchia ceylanica*; Apocynaceae; leaves; Aspidosperma alkaloids; (19*R*)-epimisiline; (19*S*)-epimisiline;  $^{13}\text{C}$  NMR.

**Abstract**—(19*R*)-Epimisiline and (19*S*)-epimisiline, two new Aspidosperma alkaloids, have been isolated from the leaves of *Petchia ceylanica*. The structures have been deduced on the basis of chemical and spectroscopic data.

### INTRODUCTION

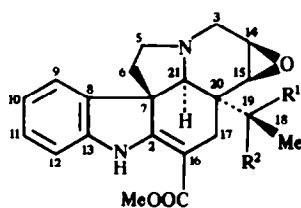
*Petchia ceylanica* Wight is an evergreen herb, indigenous to the lowlands of Sri Lanka. In this communication we report the isolation and structure elucidation of two new Aspidosperma alkaloids, (19*R*)-epimisiline (1) and (19*S*)-epimisiline (2).

### RESULTS AND DISCUSSION

(19*R*)-Epimisiline (1) was found to have the molecular formula  $\text{C}_{21}\text{H}_{24}\text{N}_2\text{O}_4$  (HRMS  $m/z$  368.1741  $[\text{M}]^+$ ). Its UV spectrum was characteristic of an anilinoacrylate chromophore showing absorptions at  $\lambda_{\text{max}}$  (MeOH) 328, 297, 226, 205 nm,  $\lambda_{\text{min}}$  (MeOH) 216, 257, 306 nm, and its IR spectrum gave absorptions at 3500 (sec. OH), 3350 (NH) and 1680  $\text{cm}^{-1}$  ( $\alpha,\beta$ -unsat. ester C=O). Its fragmen-

tation pattern indicated the presence of an aspidosperma skeleton. Exact mass measurements made on two of the important fragment ions ( $m/z$  214.0868 and 154.0868) showed them to have the molecular formulae  $\text{C}_{13}\text{H}_{12}\text{NO}_2$  (3) and  $\text{C}_8\text{H}_{12}\text{NO}_2$  (4), respectively. The formation of fragment 4 suggested that both oxygen atoms were attached to the piperidine ring.

The  $^1\text{H}$  NMR spectrum (Table 1) of (19*R*)-epimisiline showed a 3H doublet at  $\delta$ 1.15 which was assigned to the 18-methyl group ( $J_{18,19} = 7.0$  Hz), its chemical shift being consistent with the presence of a  $-\text{CH}(\text{OH})\text{CH}_3$  moiety, as in cathovoline [1] or scholaricine [2]. The  $3\alpha$  proton resonated at  $\delta$ 2.90 as a multiplet while a double doublet at  $\delta$ 3.52 ( $J_{3\beta,3\alpha} = 12.7$ ,  $J_{3\beta,14\alpha} = 5.4$  Hz) was assigned to the  $3\beta$  proton. Irradiation at  $\delta$ 2.90 caused the double doublet at  $\delta$ 3.52 for the  $3\beta$  proton to collapse into a doublet ( $J_{3\beta,14\alpha} = 5.4$  Hz).



- 1  $\text{R}^1 = \text{H}$ ,  $\text{R}^2 = \text{OH}$   
2  $\text{R}^1 = \text{OH}$ ,  $\text{R}^2 = \text{H}$

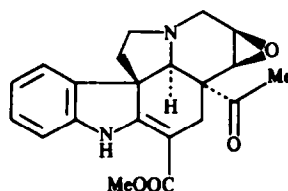
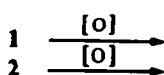


Table 1.  $^1\text{H}$ NMR data for compounds 1 and 2 (300 MHz,  $\text{CDCl}_3$ , TMS as internal standard)

H	1	2
1	8.88 s	8.87 s
3 $\alpha$ } 15 $\alpha$ }	2.90 m	2.90 m
3 $\beta$	3.52 dd	3.51 dd
6 $\beta$ } 5 $\alpha$ }	1.98 m	1.96 m
5 $\beta$	2.51 m	2.49 m
6 $\alpha$	1.76 dd	1.73 dd
9 } 10 } 11 }	6.79–7.22 m	6.78–7.16 m
14 $\alpha$	3.44 t	3.39 t
17	2.52 m	2.50 m
18	1.15 d	1.11 d
19	3.35 m	3.59 m
21 $\alpha$	3.17 s	3.17 s
OCH <sub>3</sub>	3.79 s	3.79 s

$J$  (Hz): 3 $\beta$ ,3 $\alpha$  = 12.7; 3 $\beta$ ,14 $\alpha$  = 5.4;  
6 $\alpha$ ,6 $\beta$  = 11.5; 6 $\alpha$ ,5 $\alpha$  = 4.5; 14 $\alpha$  = 4.8 (1)  
or 4.6 (2); 18,19 = 7.0.

Table 2.  $^{13}\text{C}$ NMR data for compounds 1 and 2 (75 MHz,  $\text{CDCl}_3$ , TMS as internal standard)

C	1	2
2	167.61	168.65
3	49.96	50.74
5	50.70	50.06
6	44.90	45.36
7	55.52	55.56
8	137.17	137.13
9	121.58	121.21
10	121.22	120.91
11	127.80	127.91
12	109.51	109.55
13	142.82	143.14
14	51.22	51.06
15	56.14	56.17
16	89.96	90.61
17	24.80	23.55
18	18.40	18.90
19	62.62	63.75
20	44.11	44.85
21	69.29	69.06
OCH <sub>3</sub>	51.16	51.22
C=O	168.50	168.42

The 19-methine proton geminal to the hydroxyl group resonated as a multiplet centred at  $\delta$ 3.35. Irradiation of the methyl protons resulted in the collapse of the multiplet at  $\delta$ 3.35 into a doublet ( $J_{19,\text{OH}}$  = 9.5 Hz). In the same way the doublet at  $\delta$ 1.15 for the methyl protons collapsed into a singlet when the C-19 methine proton was irradiated. A three proton singlet at  $\delta$ 3.79 was assigned to the ester methyl group. The NH proton appeared as a singlet at  $\delta$ 8.88. Each proton in the  $^1\text{H}$ NMR spectrum was individually identified by a series of 2D- $J$  resolved [3], COSY-45 [4] and homodecoupling experiments.

The  $^{13}\text{C}$ NMR spectrum (Table 2) of (19*R*)-epimisiline showed 21 carbon resonances. The multiplicity assignments were made on the basis of polarization transfer experiments (DEPT) [5]. The signal at  $\delta$ 62.62 was assigned to the hydroxyl-bearing C-19 atom. Signals for the oxymethine carbon atoms (C-14 and C-15) resonated at  $\delta$ 51.22 and  $\delta$ 56.14, respectively, the chemical shifts being typical for carbons bearing an epoxide function [6]. The presence of an epoxide unit attached to the C-14 and C-15 atoms caused some shielding of the C-17 atom due to a  $\gamma$  effect but caused little perturbation of C-21, suggesting that the epoxide was oriented *trans* to the C-20/C-21 bond. The 18-methyl carbon was found to resonate at  $\delta$ 18.4, which was 11.0 ppm downfield from its position in lochnericine [7] because of the deshielding effect of the hydroxyl group. The characteristic  $^{13}\text{C}$ NMR values confirmed the presence of the anilinoacrylate system in the molecule.

The second alkaloid (19*S*)-epimisiline (2) had a molecular formula of  $\text{C}_{21}\text{H}_{24}\text{N}_2\text{O}_4$  (HRMS  $m/z$  368.1743  $[\text{M}]^+$ ). Its mass spectrum and UV and IR spectra were almost identical to those of (19*R*)-epimisiline. Its  $^1\text{H}$ NMR spectrum corresponded closely to that of (19*R*)-epimisiline, the major differences appearing at the chemi-

cal shifts for the C-19 hydrogen atoms (Table 1).

This suggested that (19*S*)-epimisiline was the C-19 epimer of (19*R*)-epimisiline. The  $^{13}\text{C}$ NMR spectrum of (19*R*)-epimisiline was also very similar to that of (19*S*)-epimisiline. Particularly revealing was the fact that the chemical shift of C-21 in both compounds was virtually identical, the carbon resonating at  $\delta$ 69.29 in (19*R*)-epimisiline and  $\delta$ 69.06 in (19*S*)-epimisiline. It has been previously shown that when the C-14/C-15 epoxide has an  $\alpha$ -configuration, C-21 resonates at  $\delta$ 70.9 ppm,  $\delta$ 3.5 ppm downfield in comparison to the corresponding isomer in which the C-14/C-15 epoxide has a  $\beta$ -configuration [7]. This supported the conclusion that the epoxide function was in the  $\beta$  configuration in both (19*R*)-epimisiline and (19*S*)-epimisiline, since in the  $\alpha$ -configuration C-21 would have been expected to resonate upfield [7]. These results indicated that the only point of structural difference between (19*R*)-epimisiline and (19*S*)-epimisiline lay in the stereochemistry of the C-19 OH group. In order to prove this, two series of experiments were carried out. Oxidation of (19*R*)-epimisiline and (19*S*)-epimisiline with methanesulphonic anhydride/HMPA/DMSO afforded the same ketone (5) [8]. This established that the only difference in the structure of the two alkaloids was in the stereochemistry of the hydroxyl group at C-19. Acylation of each of the two alkaloids was carried out with a racemic mixture of 2-phenylbutanoic anhydride in pyridine. Recovery of the 2-phenylbutanoic acid formed during the reaction and measurement of the optical rotation of its solution gave a positive optical rotation indicating the presence of *S*-2-phenylbutanoic acid. Similarly acylation of (19*S*)-epimisiline and recovery of 2-phenylbutanoic acid produced followed by measurement of its optical rotation showed the presence of *R*-2-phenylbutanoic acid from its negative optical rotation. This established that the OH

group at C-19 in (19*R*)-epimisiline possessed an *R* configuration while the OH group in epimisiline had an *S* configuration [9]. The absolute configurations of 1 and 2 reflect the high negative optical rotations (1,  $[\alpha]_D^{27} - 382^\circ$ ; 2  $- 399^\circ$ ).

### EXPERIMENTAL

*Plant material* was collected from the Kalutara district of Sri Lanka and was identified by Prof. S. Balasubramaniam, University of Peradeniya, Peradeniya, Sri Lanka.

*Extraction and isolation.* Powdered leaves (27 kg) were extracted with MeOH (75 l). The alcoholic extract was concentrated by evaporation under reduced pressure at 40° to yield 5.4 kg of a crude concentrate which was dissolved in 5% HCl (10 l). The acid soln was extracted with CHCl<sub>3</sub> (10 l), its pH adjusted to 9 (NH<sub>3</sub>) and re-extracted with CHCl<sub>3</sub> (30 l). The latter extract was dried (Na<sub>2</sub>SO<sub>4</sub>) and after removal of the solvent the crude alkaloid mixture (140 g) was fractionated by flash chromatography over silica gel (75–230 mesh, 2 kg). Elution was carried out with petrol, petrol–CHCl<sub>3</sub> mixtures, CHCl<sub>3</sub>, CHCl<sub>3</sub>–MeOH mixtures and MeOH. The fraction obtained on elution with CHCl<sub>3</sub>–MeOH (17:3) (50 g) was rechromatographed on a column packed with silica gel (75–230 mesh), elution being carried out with CHCl<sub>3</sub> (500 ml), CHCl<sub>3</sub>–EtOAc (500 ml), EtOAc (500 ml), EtOAc–MeOH (500 ml) and MeOH (500 ml). The fraction obtained on elution with EtOAc–MeOH (9:1) (1.5 g) was subjected to TLC (silica gel) using petrol–Me<sub>2</sub>CO (4:1) to afford two new alkaloids, (19*R*)-epimisiline (1) (20 mg) and (19*S*)-epimisiline (2) (15 mg).

*Determination of C-19 configuration.* (19*R*)-Epimisiline (3.5 mg, ca 0.01 mmol) was added to a soln of racemic 2-phenylbutanoic anhydride (6.2 mg, ca 0.02 mmol) in dry C<sub>2</sub>H<sub>5</sub>N (ca 0.1 ml). The resulting mixture was allowed to stand for 16 hr at 22°; H<sub>2</sub>O (0.3 ml) was added and the mixture allowed to stand for 30 min. 0.1 M NaOH was added dropwise until the pH became 9 and the soln was then extracted with CHCl<sub>3</sub> (2 × 5 ml). The aq. layer was acidified to pH 3 using 1 M HCl and the acidic layer extracted with C<sub>6</sub>H<sub>6</sub> (2 × 5 ml). The C<sub>6</sub>H<sub>6</sub> extracts were dried (Na<sub>2</sub>SO<sub>4</sub>) filtered and the vol. adjusted to 1 ml. The optical rotation of 2-phenylbutanoic acid in the solution was found to be (+96.5) (corresponding to the *S* isomer).

An identical experiment was conducted with (19*S*)-epimisiline. The optical rotation of the soln of 2-phenylbutanoic acid in the C<sub>6</sub>H<sub>6</sub> extracts was found to be (–) (corresponding to the *R* isomer).

*(19*S*)-Oxidation of (19*R*)-epimisiline (1) and (19*S*)-epimisiline (2).* A soln of (19*R*)-epimisiline (1) (10 mg, ca 0.03 mmol) in 60 μl HMPA and 40 μl DMSO was cooled to –20° and methanesulphonic anhydride (10 mg, ca 0.06 mmol) was added. The soln was allowed to stand for 4 hr at –20°; triethylamine (ca 0.1 mmol) was added and the mixture allowed to stand at 20°

for 10 min. The soln was then poured into ice-cold water and the aq. acidic layer was basified with NH<sub>3</sub> and extracted with CHCl<sub>3</sub> (3 × 5 ml). The CHCl<sub>3</sub> extracts were dried (Na<sub>2</sub>SO<sub>4</sub>) and evaporated to yield the corresponding ketone (ca 2 mg); (19*S*)-epimisiline (2) (10 mg, ca 0.03 mmol) was oxidized under identical conditions to afford the same ketone (ca 2 mg). The identity of the ketone was confirmed by co-chromatography and comparison of spectral data.

*(19*R*)-Epimisiline (1).* Mp 252° (decomp.);  $[\alpha]_D^{27} - 382^\circ$  (c 0.3 in CHCl<sub>3</sub>); UV  $\lambda_{\text{max}}^{\text{MeOH}}$  nm: 205, 226, 297, 328 nm; UV  $\lambda_{\text{min}}^{\text{MeOH}}$  nm: 216, 257, 306; IR  $\nu_{\text{max}}^{\text{CHCl}_3}$  cm<sup>–1</sup>: 3500 (OH), 3350 (NH), 1680 (C=O); MS *m/z* (rel. int.): 368 [M]<sup>+</sup> (6), 367 (24), 350 (46), 291 (6), 214 (40), 167 (27), 154 (100), 139 (13); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>) and <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>): see Tables 1 and 2.

*(19*S*)-Epimisiline (2).* Mp 198° (decomp.);  $[\alpha]_D^{27} - 399^\circ$  (c 0.2 in CHCl<sub>3</sub>); UV  $\lambda_{\text{max}}^{\text{MeOH}}$  nm: 205, 225, 297, 327 nm; UV  $\lambda_{\text{min}}^{\text{MeOH}}$  nm: 217, 258, 306 nm; IR  $\nu_{\text{max}}^{\text{CHCl}_3}$  cm<sup>–1</sup>: 3500 (OH), 3350 (NH), 1675 (C=O); MS *m/z* (rel. int.): 368 [M]<sup>+</sup> (3), 367 (11), 350 (9), 291 (5), 214 (39), 167 (23), 154 (100), 139 (12); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>) and <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>): see Tables 1 and 2.

*Ketone (5).* UV  $\lambda_{\text{max}}^{\text{MeOH}}$  nm: 232, 296, 328 nm; UV  $\lambda_{\text{min}}^{\text{MeOH}}$  nm: 266, 306 nm; MS *m/z* (rel. int.): 366 [M]<sup>+</sup> (5), 214 (15), 179 (30), 152 (10), 135 (100), 92 (38).

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